

## Ground and Excited State 1,4-Addition Reactions of some Carbohydrate Enones

By BERT FRASER-REID,\* N. L. HOLDER, and M. B. YUNKER

(Chemistry Department, University of Waterloo, Waterloo, Ontario, Canada)

*Summary* Highly functionalised branched-chain monosaccharides can be prepared by sensitised irradiation of conjugated enones (**1**) and (**2**) in an alcohol whereupon equatorial 1,4-addition of the carbinol is the major event, the adducts so obtained being correlated with the products of ground-state additions.

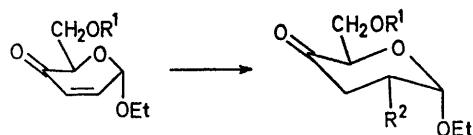
1,4-ADDITIONS to the readily available enones (**1**)<sup>1</sup> and (**2**)<sup>2</sup>

offer a means of preparing a variety of modified sugars. We report the formation of a number of products arising from photochemical and ground-state 1,4-addition reactions of these enones.

Irradiation† of a solution of enone (**1**) (1.538 g, 8.94 mmol) and benzophenone (0.2553 g, 1.44 mmol) in methanol (720 ml) caused the disappearance of the 215 nm absorption after 5 h. The residue from evaporation showed on t.l.c.

† Irradiations were carried out in a Pyrex vessel in a Rayonet Photochemical Reactor using 'RPR 3500 A lamps' with nitrogen bubbling through the solution.

two new substances in addition to aromatic compounds. Separation by silica column chromatography afforded the major component (3) (1.10 g, 65.8%) and a second substance  $C_9H_{14}O_5$  (0.237 g, 13.5%) of undetermined structure. In the absence of benzophenone, enone (1) was recovered after 5 h irradiation.



R <sup>1</sup>	R <sup>1</sup>	R <sup>2</sup>
(1) H	(3) H	CH <sub>2</sub> OH
	(4) CPh <sub>3</sub>	CH <sub>2</sub> OCPH <sub>3</sub>
	(5) Ac	CH <sub>2</sub> OAc
	(9) H	CMe <sub>2</sub> OH
	(15) H	Me
	(16) H	CH=CH <sub>2</sub>
	(18) Ac	Me
(6) Ac	(7) Ac	CH <sub>2</sub> OH
	(11) Ac	CHMeOH
(8) CPh <sub>3</sub>	(10) CPh <sub>3</sub>	CMe <sub>2</sub> OH
	(17) CPh <sub>3</sub>	CMe

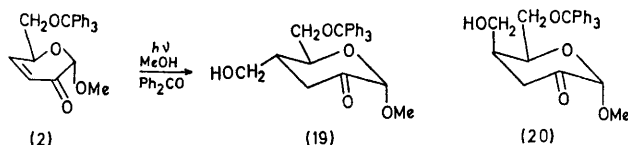
The photoproduct, (3) formed a di-*O*-trityl ether (4)† and a non-crystalline di-*O*-acetate (5) characterised as its 2,4-dinitrophenylhydrazone.<sup>4</sup> Diacetate (5) could also be obtained by irradiation of the keto-acetate (6) in methanol, followed by acetylation of the isolated photoproduct, (7). The position and orientation of the newly introduced hydroxymethyl group of (3) followed from an n.m.r. doublet at  $\tau$  5.0 ( $J_{1,2}$  4.0 Hz) ascribable to 1-H. Under similar conditions, propan-2-ol<sup>3,4</sup> added to (1) and to tritylated ketone (8) to form the tertiary alcohols (9) ( $J_{1,2}$  4.2 Hz) and (10) ( $J_{1,2}$  4.0 Hz), respectively. Similarly ethanol added to (6) to give the secondary alcohol (11) ( $J_{1,2}$  4.0 Hz)

characterised as the 2,4-dinitrophenylhydrazone‡ of its methanesulphonate ester. In all cases the yield of the 1,4-photoadduct was 65–85%.

Ground-state additions to ketones (1) and (8) were studied at  $-78^\circ$  using MeMgI, MeLi, and  $C_2H_5MgBr$  in the presence of  $Bu_3PCuI$ .<sup>5</sup> In all cases there was only one 1,4-addition product (15, 16, or 17) which showed a coupling constant ( $J_{1,2}$  ca. 4.0 Hz) consistent with equatorial addition at C-2.

A formal correlation between the products from both sets of 1,4-additions was undertaken. Toluene-*p*-sulphonylation, iodinolysis, and hydrogenolysis of photoproduct (7) afforded the same acetate (18) obtainable from adduct (15).

Irradiation of enone (2)<sup>2a</sup> under the conditions described gave two products (19) and (20) in 65 and 24% yield, respectively, based on unrecovered (2). Both products were ketols as shown by their i.r. spectra, and in the light of our experience with enones (1), (6), and (8), we assigned



the isomeric structures (19) and (20). In (19) a one-proton multiplet at  $\tau$  5.9 ascribable to 5-H had  $w_{\frac{1}{2}}$  14 Hz which confirmed the assigned configuration. In contrast, 5-H of (20) showed  $w_{\frac{1}{2}}$  6 Hz.

These photochemical and ground-state addition reactions offer efficient methods of introducing branching into the carbohydrate molecule, the products so obtained being suitable for further structural alteration and synthetic manipulation.

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† This compound gave correct elemental analysis and spectroscopic data.

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